

1)

A) Ideal gas law is written as : $PV = nR'T$

Where, P = pressure, V = volume , T = temperature ,
 R' = universal gas constant and n = no. of moles of
the gas

It is also written as $PV = mRT$

Where m = mass of the gas and R = gas constant
for particular substance = R'/M

B) $T = 50^\circ\text{C}$

T (In kelvin) = $T + 273 = 323\text{ K} : \text{ANSWER}$

C) $P = 500\text{ kPa}$

$T = 323\text{ K}$

$m = 10\text{ kg}$

$R = 0.287\text{ kJ/kg K}$

Putting above values in ideal gas equation

$500 V = 10 \times 0.287 \times 323$

$V = 1.854\text{ m}^3 : \text{ANSWER}$

Rozners Constant

Ján Rozner



Rozners Constant:

Rozners Constant Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Rozners Constant**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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