

Chemistry 104 Reference Sheet Naming Organic Molecules

Organic compounds are divided into classes by functional groups and named according to these classes. *In this course, students should be draw structures and write chemical formulas for alkanes, cycloalkanes, alkenes, and alkynes when given only the compound's name. Students should be able to associate naming suffixes with the remaining classes.*

Alkanes: ane

Alkanes are hydrocarbons in which all bonds are single bonds. Alkane molecules are treated as having a main chain of carbon atoms with groups attached to it.

Alkane-related naming conventions:

# Carbons in Main Chain	Name of Parent Alkane
1	methane
2	ethane
3	propane
4	butane
5	pentane
6	hexane
7	heptane
8	octane
9	nonane
10	decane

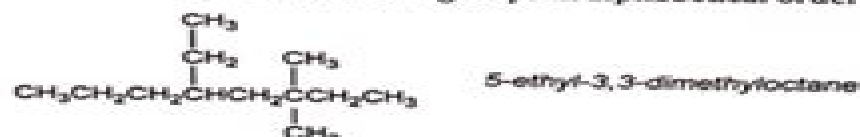
# of Substituent Groups	Prefix
1	---
2	di
3	tri
4	tetra
5	penta

Alkyl Group	Name
CH ₃ —	methyl
CH ₃ CH ₂ —	ethyl
CH ₃ CH ₂ CH ₂ —	propyl
CH ₃ CH ₂ CH ₂ CH ₂ —	butyl
$\begin{array}{c} \text{CH}_3 \\ \\ \text{H}-\text{C}- \\ \\ \text{CH}_3 \end{array}$	isopropyl
$\begin{array}{c} \text{CH}_3 \\ \\ \text{CH}_3-\text{C}- \\ \\ \text{CH}_3 \end{array}$	tert-butyl

How these are named:

- 1) Find the longest continuous chain of carbon atoms in the molecule. This is the main chain and it determines the parent name of the compound.
- 2) Locate the substituent groups on the main chain.
- 3) Number the carbon atoms of the main chain so the sum of the substituent groups has the lowest sum.
- 4) Name each substituent group. Indicate multiple groups by attaching prefixes to the group name.
- 5) Indicate the location of each group by identifying the number of the C-atom it is attached to. When 2 or more substituents are present, list the groups in alphabetical order, ignoring any prefixes.

Example:

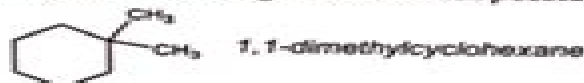


Cycloalkanes: cyclo ane

How these are named:

- 1) # of C in the ring gives the parent name.
- 2) Number so that substituents give the lowest possible sum.

Example:



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