

# HOW TO Build Your Child's Self-Esteem

By the Kids in Transition to School (KITS) Program



**Use praise**  
that is specific,  
sincere & based  
on efforts (vs  
outcomes)



Let your child  
make her  
**own  
decisions**  
when possible



Let your child  
take  
**reasonable  
risks**  
to build self-  
confidence



teach your child  
**positive self-  
talk**



help your child to  
**learn from  
mistakes**  
and problem solve



give your child  
**responsibilities**  
to show him what  
he can do



let your child  
**take pride in  
accomplishments**



be a positive  
self-esteem  
**role model**



with positive self  
esteem your child  
will have more ...

**confidence**

**willingness to try  
new things**

**perseverance in the  
face of challenges**

# Me A Curriculum For Teaching Selfesteem In The Classroom

**Thomas Griffiths**



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web during oxidation and reduction the oxidation numbers of atoms change if an atom is oxidized its oxidation number increases ie it becomes more ve or less ve if an atom is reduced its oxidation number decreases ie it becomes less ve or more ve

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web write the oxidation and reduction reactions for each redox reaction the first one is done for you  $23 \text{ Fe} + \text{Co} \rightarrow \text{Co}^{2+} + \text{Fe}$   
oxidation  $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^-$  reduction  $\text{Fe}^{2+} + 2\text{e}^- \rightarrow \text{Fe}$   $24 \text{ 3 Ag} + \text{Ni} \rightarrow \text{Ni}^{3+} + 3 \text{ Ag}$  oxidation reduction  $25 \text{ Cu} + \text{Pb} \rightarrow \text{Pb}^{2+} + \text{Cu}$  oxidation reduction  $26 \text{ O}_2$

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web in each of the following redox reactions indicate the elements that undergo changes in oxidation number and the magnitude of change then identify the element oxidized and reduced as well as the oxidizing agent and reducing agent  $\text{a } \text{I}_2\text{O}_5$   
 $5\text{CO} + \text{I}_2\text{O}_5 \rightarrow 5\text{CO}_2 + \text{I}_2$   $5\text{CO} + \text{I}_2\text{O}_5 \rightarrow 5\text{CO}_2 + \text{I}_2$

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