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Micelles: Self-organized surfactant assemblies

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Surfactants can self-organize under specific environmental conditions in solution to form 'micelles'. The micelles are of two types, 'normal micelles' (called micelles), which are formed by surfactant association in water or polar solvents, and 'reverse micelles', which are formed in nonpolar media. The surfactants and micellar solutions have versatile uses in the fields of chemistry, biochemistry, pharmacy, medicine and industry to augment and control solubilization, stabilization, dispersion, cleaning, rates of chemical reactions, enhanced oil recovery, etc. In this review, fundamentals of formation of micelles, their physico-chemical properties and probable uses are presented for an overall grasp of the topic of contemporary interest.

Ampurements are chemical compounds having dual affinity for water and oil. They have distinct nonpolar (liophilic or hydrophobic) and polar (hydrophilic or liophobic) sections in their molecules. Soaps, detergents, long chain alcohols (amines, aldehydes, etc.) and lipids constitute the class of amphiphiles. They are surface active, can reduce surface tension of the medium or interfacial tension between two immiscible liquids (e.g. oil and water), can assist solubilization, cleaning, dispersion, enulsification, etc. The amphiphiles comprising soaps and detergents show a special property in solution manifesting characteristic self-organization or association

called 'micelle' formation12. Under appropriate conditions, amphiphiles (mostly of lipid types) may also form 'liquid crystals' in solution. The amphiphiles that form micelles and can be potentially used for surface chemical works are termed SURFace ACTive AgeNTS or SURFACTANTS. Soaps and detergents come under this heading. Soaps are prepared by the hydrolysis (saponification) of naturally occurring fats and oils, and detergents are synthetically prepared. Big industries are busy with the production of soaps and detergents all over the world. The innumerable possibilities of such products through chemical modification and synthesis and great many uses of amphiphiles in pharmaceutical, chemical, biochemical (including biomedical) and industrial fields have added tremendous scope of research and study in this area of surface and colloid science and technology. The investigations on micelles comprise a significant share in this area, a comprehensive accounting of which is of contemporary interest. In the following sections, the overall state of the art of basic aspects of surfactants and micelles including their potential uses are presented.

Types of micelle-forming amphiphiles or surfactants

The micelle-forming amphiphiles or surfactants essentially fall in two categories, 'ionic' and 'nonionic'. They Organized Assemblies In Chemical Analysis Vol 1
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