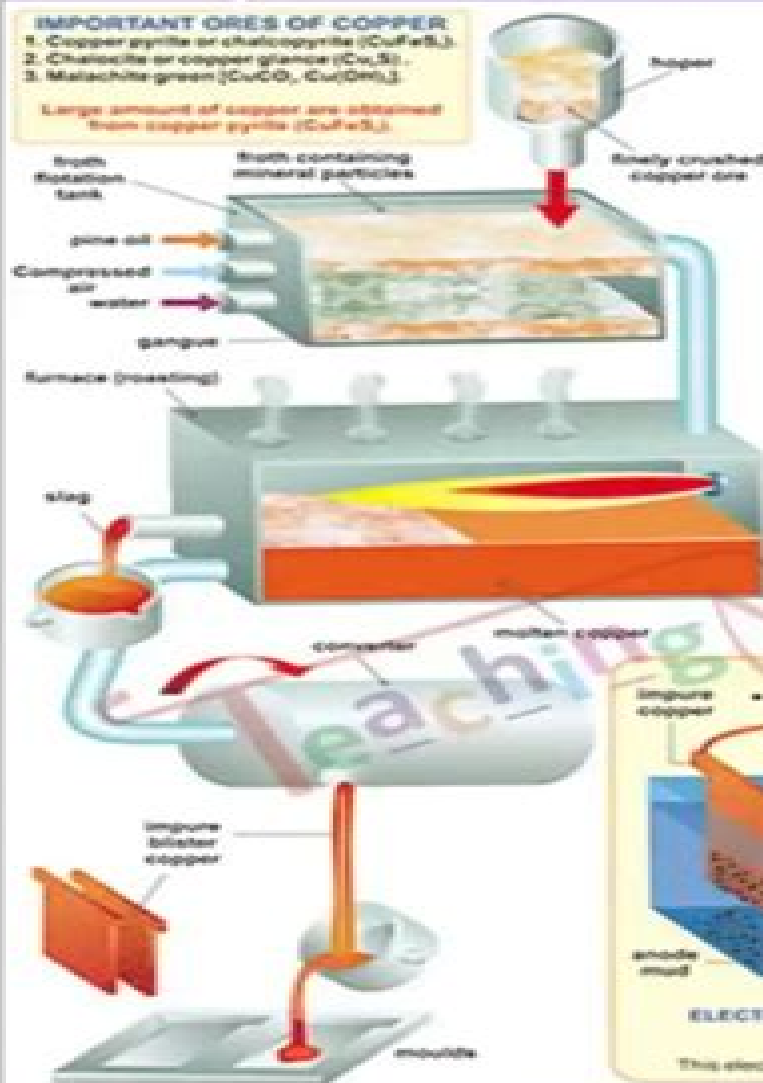


Copper Metallurgy

IMPORTANT ORES OF COPPER

1. Copper pyrite or chalcopyrite (CuFeS_2)
2. Chalcocite or copper glance (Cu_2S)
3. Malachite-green ($\text{CuCO}_3 \cdot \text{Cu(OH)}_2$)

Large amount of copper are obtained from copper pyrite (CuFeS_2).



CONCENTRATION

The finely crushed ore is concentrated by Froth-Flotation process.

ROASTING

Sulfur is oxidized to SO_2 and impurities of arsenic and antimony are removed as volatile oxides. The following reaction takes place:



Cuprous sulphide and ferrous sulphide are further oxidized into their oxides.



SMELTING

The roasted ore is mixed with coke and silica (sand) SiO_2 and is introduced into a blast furnace. The top air is blasted and FeO is converted into ferrous silicate (FeSiO_3).



FeSiO_3 (slag) floats over the molten matte of copper.

BESSEMERIZATION

Copper metal is extracted from molten matte through bessemerization. The matte is introduced into Bessemer converter. The air blown through molten matte. Blast of air converts Cu_2S partly into Cu_2O which reacts with remaining Cu_2S to give molten copper.

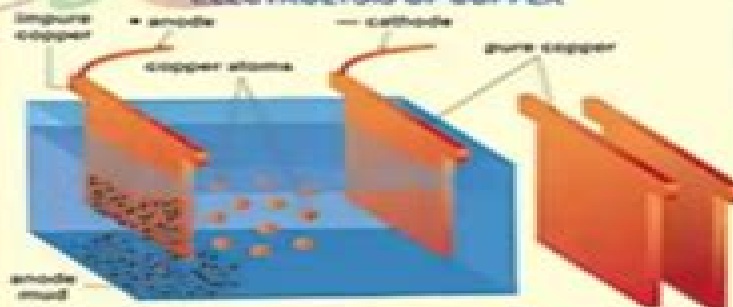


The copper so obtained is called 'Blister Copper' as escaping SO_2 forms blister like formations.

REFINING OF COPPER

Blister copper is refined by electrolysis.

ELECTROLYSIS OF COPPER



ELECTROCHEMICAL CHANGES DURING ELECTROLYSIS



This electrically refined copper is 100% pure.

Uses of Copper

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Table of Contents Metallurgy Of Copper

1. Understanding the eBook Metallurgy Of Copper
 - The Rise of Digital Reading Metallurgy Of Copper
 - Advantages of eBooks Over Traditional Books
2. Identifying Metallurgy Of Copper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Metallurgy Of Copper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Metallurgy Of Copper
 - Personalized Recommendations
 - Metallurgy Of Copper User Reviews and Ratings
 - Metallurgy Of Copper and Bestseller Lists
5. Accessing Metallurgy Of Copper Free and Paid eBooks
 - Metallurgy Of Copper Public Domain eBooks
 - Metallurgy Of Copper eBook Subscription Services

- Metallurgy Of Copper Budget-Friendly Options
- 6. Navigating Metallurgy Of Copper eBook Formats
 - ePub, PDF, MOBI, and More
 - Metallurgy Of Copper Compatibility with Devices
 - Metallurgy Of Copper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Metallurgy Of Copper
 - Highlighting and Note-Taking Metallurgy Of Copper
 - Interactive Elements Metallurgy Of Copper
- 8. Staying Engaged with Metallurgy Of Copper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Metallurgy Of Copper
- 9. Balancing eBooks and Physical Books Metallurgy Of Copper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Metallurgy Of Copper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Metallurgy Of Copper
 - Setting Reading Goals Metallurgy Of Copper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Metallurgy Of Copper
 - Fact-Checking eBook Content of Metallurgy Of Copper
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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