



Resolving a Two-Component Mixture

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Purpose of the Experiment

Separate and recover the components of a mixture of sand and sodium chloride of unknown proportions. Calculate the percent of each component in the mixture and the percent recovery of the components.

Background Required

You should be familiar with basic techniques for measuring volume and mass. You should understand the concepts associated with solution chemistry and stoichiometry.

Background Information

A **binary mixture** is a mixture containing only two components. We can use a variety of methods, either physical or chemical, to separate the components of such a mixture. **Physical separation methods** are based on differences between the physical properties of the components, such as solubility or boiling point.

One physical method for separating a liquid from a solid is **decantation**. First, we allow the solid to settle to the bottom of the container. Then we carefully pour off all of the liquid, called the **supernatant liquid**, or **supernate**, into another container, without disturbing the solid.

We can also physically separate a solid from a liquid by **filtration**, a method that involves pouring the mixture onto a porous material, such as filter paper. The solid, called the **residue**, is unable to pass through the pores in the filter paper, so it is retained on the paper. The liquid, which passes through the paper, is called the **filtrate**.

Evaporation is a third physical separation method, in which a solution is heated to vaporize and thus remove the solvent. The solid remaining after we have evaporated the solvent was the solute.

When one component of a binary mixture of two solids is soluble in a particular solvent, and the other component is not, we can separate the components using a fourth physical method, **extraction**. We add the solvent, dissolving the soluble component, then separate the mixture using filtration.

Example Problem

Determine the percent recovery of silver bromide (AgBr) and potassium bromide (KBr) when separated from a sample of a binary mixture weighing 2.18 g, according to the following procedure.

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M Lipman



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