



Reaktionen in der Organischen Chemie 1

Alkane

*nucleophile Substitutionen



primäre Substrate



sekundäre Substrate



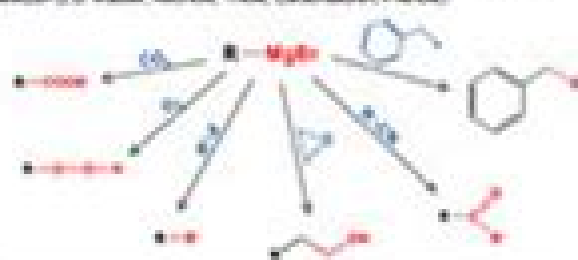
Radikale

Teilchen, die mindestens ein freies Elektronenpaar haben und oft negativ geladen sind (Anionen). Sie sind stark reaktiv und können leicht mit anderen Molekülen reagieren. Sie werden unter Verwendung eines Radikalengenerators wie Azobisisobutyronitril (AIBN) erzeugt, der bei Erwärmen in Radikale zerfällt.

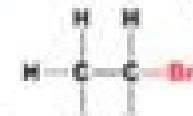
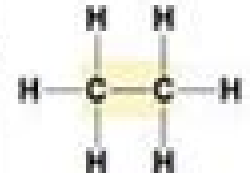
Substituierte Alkanolierung

Reaktionen von Grignard-Verbindungen

Grignard-Verbindungen reagieren mit Substraten, welche einen Protonenakzeptor (z.B. Wasser, Alkohole, Thiole, Carbonsäuren, Ester) besitzen.



Wichtig: Die nucleophile Substitution findet bevorzugt an sekundären und tertiären Kohlenstoff-Atomen statt. Primäre Kohlenstoffe sind weniger reaktiv.



Halogenierung (radikalische Substitution)

Grignard-Synthese

Mercaptan-Synthese*

Hydroxy-Substitution*

Finkelstein-Reaktion* (halogenatom-Transfer)

Ammonolyse*

Kolbe-Nitrilsynthese*

als Nebenprodukt entstehen Isocyanide R-N=C

Wurtz-Fittig-Reaktion



Eliminierung

Williamson-Ethersynthese*



Oxidation (Verbrennung)

Die radikalische Substitution funktioniert auch mit Cl_2 und I_2 . Fluor reagiert so heftig, dass zwischen den Halogenatomen des Alkans zerlegt.

$\text{Br}_2/\text{h}\nu$ Verbindungen, die in der Reaktion umgewandelt werden können.

Reaktionsmechanismen dieser Stoffgruppen sind auf einer anderen Übersicht dargestellt.

Radikale Und Radikationen In Der Organischen Synthese

Shasha Hu

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red circular gradient shape partially visible behind it.

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