



Reaktionen in der Organischen Chemie 1

Alkane

Che

*nucleophile Substitutionen



primäre Substrate



sekundäre Substrate



Radikale

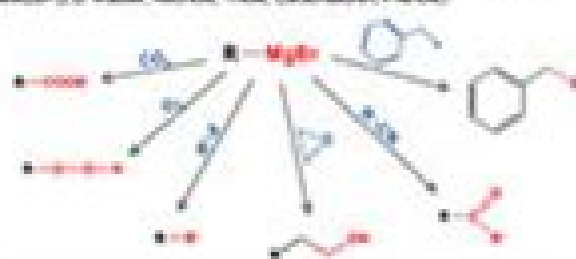
Radikale, die mindestens ein freies Elektronenpaar haben und oft negativ geladen sind (anionisch) bzw. eine starke negative Partialladung besitzen. Sie treten unter Verwendung eines Radikalengenerators wie katalytische Wirkung in einem geschlossenen System auf.

Substitution

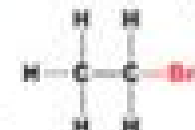
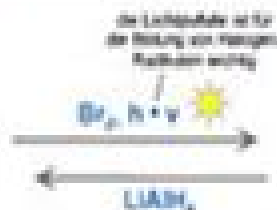
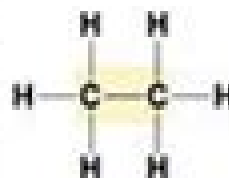
Radikalische

Reaktionen von Grignard-Verbindungen

Grignard-Verbindungen reagieren mit Substraten, welche einen Protonenakzeptor (z.B. Wasser, Alkohole, Thiole, Carbonsäuren, Ester).



Wichtig: Die radikalische Substitution findet bevorzugt an sekundären und tertiären Kohlenstoff-Atomen statt (abhängig von der Substitution der Alkylkette).



Halogenierung (radikalische Substitution)

Grignard-Synthese

Mercaptan-Synthese*

Hydroxy-Substitution*

Finkelstein-Reaktion* (halogenatom-Vertauschung)

Ammonolyse*

Kolbe-Nitrilsynthese*

als Nebenprodukt entstehen Isocyanide R-N=C

Wurtz-Friedel-Crafts-Reaktion



Eliminierung

Williamson-Ethersynthese*



Oxidation (Verbrennung)

Die radikalische Substitution funktioniert auch mit Cl_2 und I_2 . Fluor reagiert so heftig, dass zwischen der Substitution und der Verbrennung keine Unterscheidung mehr möglich ist.

Radikalische Substitutionen sind in der Reaktion umgewandelt worden. Radikale.

Reaktionsmechanismen dieser Stoffgruppen sind auf einer anderen Übersicht dargestellt.

Radikale Und Radikationen In Der Organischen Synthese

G Orfield



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