

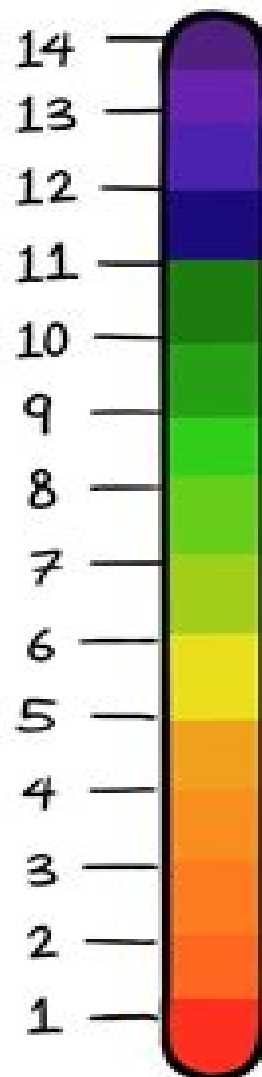
What is pH?

The **pH scale** is a measure of how **acidic** or **basic** a solution is.

Solutions with *lower* concentrations of hydrogen ions have **higher pHs** and are considered **basic**.

pH of 7 is **neutral**. At room temperature, pure water has a pH of 7.

Solutions with *higher* concentrations of hydrogen ions have **lower pHs** and are considered **acidic**.



pH is calculated by taking the negative log of the concentration of hydrogen ions $[H^+]$ in an aqueous solution.

Colors shown are for the universal indicator.

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