

Rutherford and the Nucleus of the Atom

- Rutherford used the radioactive alpha particles discovered by Becquerel to establish the nature of the nucleus of the atom
- He directed the positively charged He atoms (alpha particles) at a piece of gold foil
- Most passed through
- Some were deflected
- A few bounced back
- Rutherford was able to reason that there must be a positive portion in the center of the atom because like repels like
- This portion was referred to as the nucleus
- Most of the atom is empty space

Rutherford The Nature Of The Atom

Didier Musso



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