

pH Laboratory Activity

Name: _____

Station #	Test	Circle the Results	Circle the pH
A	Red Litmus	Red or Blue	Acid Base Neutral
	Blue Litmus	Red or Blue	
B	Red Litmus	Red or Blue	Acid Base Neutral
	Blue Litmus	Red or Blue	
C	Red Litmus	Red or Blue	Acid Base Neutral
	Blue Litmus	Red or Blue	

Station #	Test	Results	pH Value
A	Universal pH Paper	colour	pH
B	Universal pH Paper	colour	pH
C	Universal pH Paper	colour	pH

Station #	Test	pH Value
A	Normal Probe	pH
B	Normal Probe	pH

Station #4

Acid Base Neutralization Reaction of hydrogen chloride and sodium hydroxide via double displacement reaction.



+

Procedure

1. Add 10 drops of NaOH (base) on the evaporating dish.
2. Add 5 drops of phenolphthalein indicator on the evaporating dish and gently swirl the evaporating dish.
3. Slowly add 5 drops of HCl (acid) on the evaporating. Swirl the evaporating dish.
4. Slowly add another 5 drops of HCl (acid) on the evaporating. Swirl the evaporating dish. When the colour disappears, the solution has been neutralized.

Note: If the solutions are the exact same concentration, 10 drops of acid should neutralize 10 drops of the base to produce water and sodium chloride.

5. How many drops did you need to neutralize the sodium hydroxide (base)?

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Xiaolong Qi



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