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web q2 what is the disadvantage of fractional crystallization the disadvantage of fractional crystallization is that with each increase in crystallization step number we lose kno3 the graph shows that as crystallization number increases the of the original kno3 mass recovered decreases in the first

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web fractional crystallization is a method of refining substances based on differences in solubility it fractionates via differences in crystallization forming of crystals if a mixture of two or more substances in solution are allowed to crystallize for example by allowing the temperature of the solution to decrease the precipitate will

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web the percent recovery of the unknown was 78 shown in calculation 2 meaning again a significant amount of mass was lost during the recrystallization process the melting range of the unknown was observed to be 64 66 c and thus can be identified as biphenyl which has a melting range of 69 71 c

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web data calculations and graphs part i fractional crystallization of kno 3 with nh 4 2 fe so 4 2 6h 2 o impurity a starting components fract crystallization 0 step 3 mixture before first crystallization

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web fractional crystallization lab 3 scale tray tub trough ice water salt of the rock variety stirring rod funnel hot plate bunsen burner beaker tongs procedure measure the mass of the beakers and record next measure 50 ml of the nacl kno3 solution

into the graduated cylinder then pour into the beaker

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