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- **Definition:**

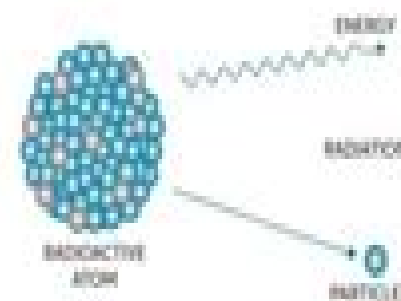
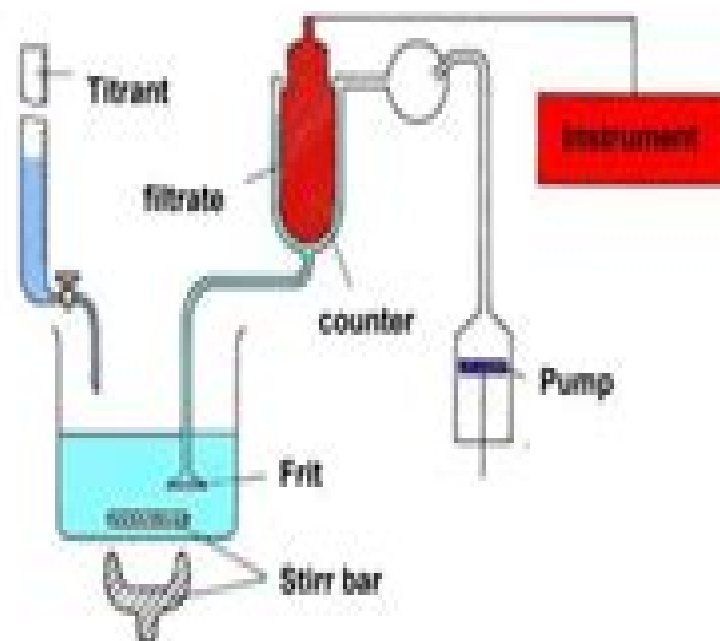
Radiometric titration is an analytical technique that uses radioactive isotopes to measure the concentration of an analyte in a sample. It is a form of titration that relies on the principle of radioactive decay.

- **Principle:**

- It is based on radioactive decay, which is the spontaneous breakdown of an unstable atomic nucleus into a more stable nucleus, emitting radiation in the process.
- The rate of radioactive decay is proportional to the number of radioactive atoms present in the sample, and this can be used to determine the concentration of the analyte being measured.

- **Radioactive Isotopes :**

- Carbon-14, Tritium.
- Phosphorus-32, Sulfur-35.
- Iodine-125



Radiometric Titrations

SJ Ball



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