

Radioisotope Techniques

Radioactivity is spontaneous disintegration of atomic nuclei. The nucleus emits α particles, β particles, or electromagnetic rays during this process. Approximately 3000 nuclides have been discovered so far that are unstable. They undergo spontaneous fission to achieve stability. During this process, the *parent nuclide* transforms into an atom of a different type, called the *daughter nuclide*.

Units of Radioactivity

Radioactivity is measured in *curie*. It is disintegration rate of 1 g radium that is 3.7×10^{10} disintegrations per second. SI unit for radioactivity is Becquerel (Bq), which is defined as one disintegration per second.

$$\begin{aligned} 1 \text{ Curie (Ci)} &= 3.7 \times 10^{10} \\ \text{disintegrations per second (dps)} & \\ &= 2.22 \times 10^{10} \text{ disintegrations per} \\ \text{minute (dpm)} & \end{aligned}$$

$$1 \text{ millicurie (mCi)} = 3.7 \times 10^7 \text{ dps}$$

$$1 \text{ microcurie (}\mu\text{Ci)} = 3.7 \times 10^4 \text{ dps}$$

$$1 \text{ Becquerel (Bq)} = 1\text{dps} = 2.7 \times 10^{-11} \text{ Ci}$$

$$1 \text{ Kilobecquerel (KBq)} = 10^3 \text{ dps} = 2.7 \times 10^{-8} \text{ Ci}$$

Radioisotope Techniques

Michael Brown



Radioisotope Techniques:

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