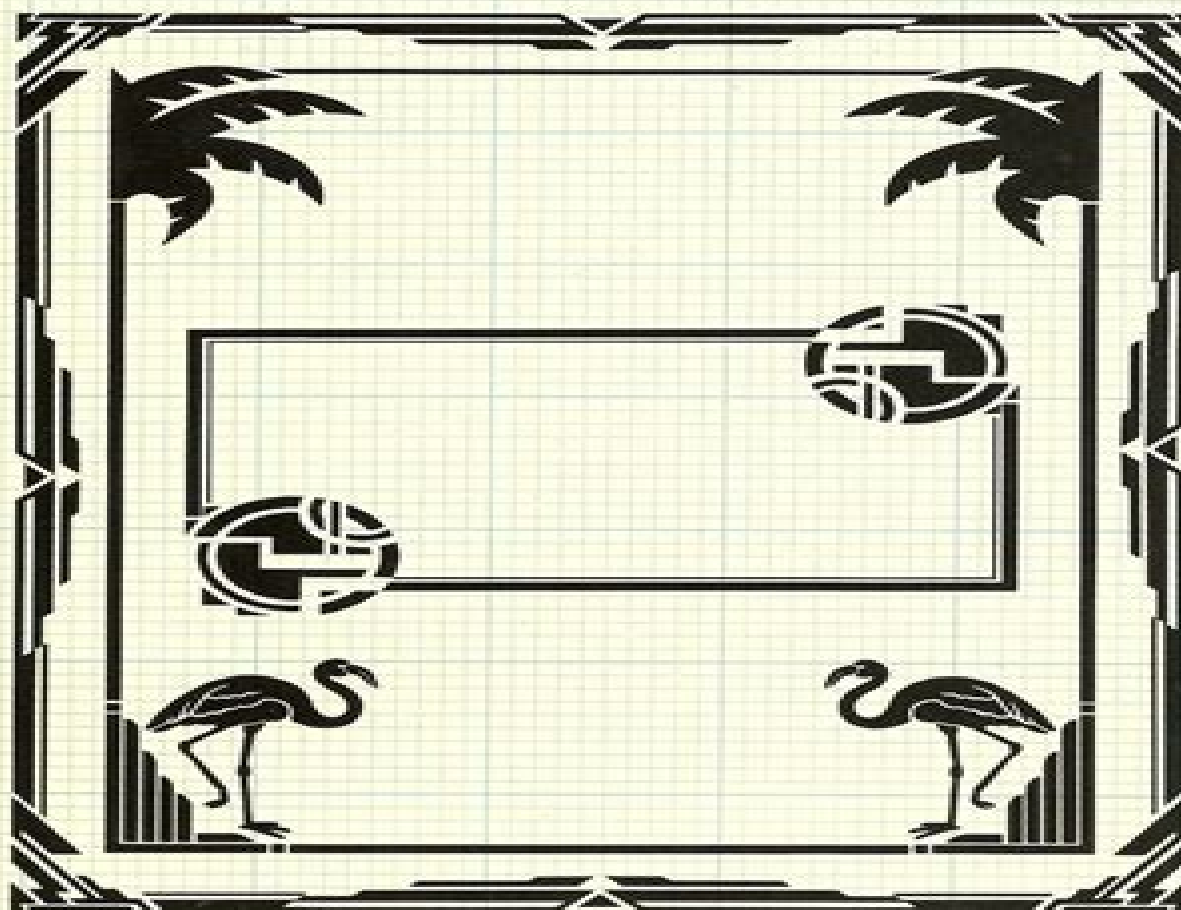


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web heat capacity is the amount of heat required to change the temperature of a given amount of matter by 1 °C the heat capacity of 1 gram of a substance is called its specific heat

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web the specific heat is numerically equal to the amount of heat necessary to change the temperature of 1.00 kg of mass by 1.00 °C the SI unit for specific heat is J/kg

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web in order to quantify the heat transfer q that is associated with changing the temperature of a substance the mass m and specific heat c of that substance must be multiplied by

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web use this calculator to determine the heat transferred to or from a substance $q = c\Delta t$ in joules j kilojoules kj british thermal units btu iso calories cal or kilocalories kcal

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conduction also known as thermal conduction is the transfer of internal energy within a material the energy is transferred by collision of molecules atoms and electrons inside the material the heat flow will occur in solid liquid and plasma phases and the energy will always flow from hot to cold joseph see more

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web if the reaction releases heat $q_{rxn} < 0$ then heat is absorbed by the calorimeter $q_{calorimeter} > 0$ and its temperature increases conversely if the reaction absorbs heat

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