PROPERTIES OF WATER

Structure



2 hydrogen atoms bond to oxygen atom, resulting in a bent shape

States of Matter



Melting = 0 °C Boiling = 100 °C Point Point

Solid (Ice) (Water)

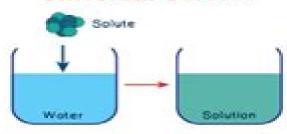
Gas (Vapor)

Density



The density is maximum (1 g/cm³) at 3.98 °C

Universal Solvent



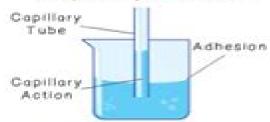
Capable of dissolving many substances

Surface Tension



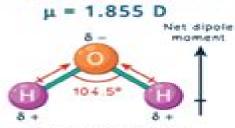
Forms a film that can resist an external force

Capillary Action



Moves up a thin air column against gravity

Polarity & Dipole Moment



Polar molecule due to the asymmetric shape

Heat of Vaporization

 $\Delta H_{\nu} = 2260 \text{ J/g}$



Specific Heat

c = 4.184 J/g -°C





Properties Of Water

Annelies Wilder-Smith

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2

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