#### CHEM 1405 Practice Exam # 3 (Ch 5-7)

Section # Name MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question. 1) The ability of an atom to attract the shared electrons in a covalent bond is its \_\_\_\_\_\_\_ 1) A) electronegativity B) nonpolarity C) ionic character D) bonding ability E) polarity Which of the following substances contains a nonpolar covalent bond? E) H<sub>2</sub>O A) NaCL B) N2 C) MgF2 Do Nilla 3) A polar covalent bond is found in which of these compounds? E) Na(C) All Ho B) H>O C) No. D9 F2 One mole of copper(II) sulfate, CuSO<sub>4</sub>, contains \_\_\_\_\_\_ O atoms. A)  $6.02 \times 10^{23}$ E)  $2.41 \times 10^{23}$ C) 4  $D32.41 \times 10^{24}$ E)  $1.51 \times 10^{23}$ 5) How many moles of carbon atoms are there in 0.500 mole of C<sub>2</sub>H<sub>6</sub>? 5) A) 0.500 mole B) 4.00 moles C1 3.00 moles D)  $6.02 \times 10^{23}$  moles E) 1.00 mole 6) What is the molar mass of sodium phosphate, Nag.PO4? 6) A) 118.0 g/mole B) 163.9 g/mole C) 226.1 g/mole D) 69.96 g/mole E) 354.0 g/mole 7) How many moles of water, H2O, are present in 75.0 g. H2O? A)  $1.35 \times 10^3$  moles B) 75.0 moles C) 4.16 modes D) 7.50 moles E) 4.41 motes

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