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web the following electron configurations belong to which elements 21 1s²2s²2p⁶3s¹ sodium 22

1s²2s²2p⁶3s²3p⁴4s²3d¹⁰4p⁶5s²4d⁶ ruthenium 23 kr 5s²4d¹⁰ cadmium 24 xe 6s²4f¹⁴5d¹⁰6p² lead 25 rn 7s²5f¹⁴6d⁴

seaborgium determine if the following electron configurations are correct 26 1s22s22p63s23p64s24d104p65s1 no it should be 3d10

3 1 electron configurations problems chemistry libretexts - Aug 19 2023

web how many protons neutrons and electrons are in atoms of these isotopes write the complete electron configuration for each isotope answer co has 27 protons 27 electrons and 33 neutrons 1s 2 2s 2 2p 6 3s 2 3p 6 4s 2 3d 7 i has 53 protons 53 electrons and 78 neutrons 1s 2 2s 2 2p 6 3s 2 3p 6 3d 10 4s 2 4p 6 4d 10 5s 2 5p 5

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web select the electron configuration tab on show number of electrons introduction beyond argon it is a bit tricky to determine which subshell gets filled next there are several rules that scientists use to determine the electron configurations of larger atoms question how are the electron configurations of elements beyond argon determined