

MODERN PHYSICAL CHEMISTRY

An Introductory Text

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A. H. Holliday

Butterworths

Modern Physical Chemistry An Introductory Text

Robert G. Mortimer



Modern Physical Chemistry An Introductory Text:

Modern physical chemistry Hermann Block, Arthur K. Holliday, 1976 *Fundamentals of Physical Chemistry* [by] H.D. Crockford & Samuel B. Knight H.D. Crockford, 1967 Modern Physical Chemistry: an Introduction Text Hermann Block, Arthur Kenneth Holliday, 1973 *Modern Physical Chemistry* G.H. Duffey, 2013-11-11 In this new textbook on physical chemistry fundamentals are introduced simply yet in more depth than is common Topics are arranged in a progressive pattern with simpler theory early and more complicated theory later General principles are induced from key experimental results Some mathematical background is supplied where it would be helpful Each chapter includes worked out examples and numerous references Extensive problems review and discussion questions are included for each chapter More detail than is common is devoted to the nature of work and heat and how they differ Introductory Caratheodory theory and the standard integrating factor for dG_{rev} are carefully developed The fundamental role played by uncertainty and symmetry in quantum mechanics is emphasized In chemical kinetics various methods for determined rate laws are presented The key mechanisms are detailed Considerable statistical mechanics and reaction rate theory are then surveyed Professor Duffey has given us a most readable easily followed text in physical chemistry **Physical Chemistry** William M. Davis, 2011-12-06 Designed specifically for a two semester introductory course sequence in physical chemistry this text presents core principles and topics Straightforward and streamlined it presents the necessary amount of detail for comprehension Organized in such a way that the various topics covered are connected to each other it allows students to see physical chemistry as an interconnected discipline and not a series of unrelated concepts Each chapter in this new edition has been thoroughly updated and includes new information on computational applications more end of chapter problems and new chapters on nanotechnology and surface chemistry *TEXTBOOK OF PHYSICAL CHEMISTRY* H. K. MOUDGIL, 2014-10-21 This comprehensive textbook now in its second edition is mainly written as per the latest syllabi of physical chemistry of all the leading universities of India as well as the new syllabus recommended by the UGC This thoroughly revised and updated edition covers the principal areas of physical chemistry such as thermodynamics quantum chemistry molecular spectroscopy chemical kinetics electrochemistry and nanotechnology In a methodical and accessible style the book discusses classical irreversible and statistical thermodynamics and statistical mechanics and describes macroscopic chemical systems steady states and thermodynamics at a molecular level It elaborates the underlying principles of quantum mechanics molecular spectroscopy X ray crystallography and solid state chemistry along with their applications The book explains various instrumentation techniques such as potentiometry polarography voltammetry conductometry and coulometry It also describes kinetics rate laws and chemical processes at the electrodes In addition the text deals with chemistry of corrosion and nanomaterials This text is primarily designed for the undergraduate and postgraduate students of chemistry B Sc and M Sc for their course in physical chemistry Key Features Gives a thorough treatment to ensure a solid grasp of the material

Presents a large number of figures and diagrams that help amplify key concepts Contains several worked out examples for better understanding of the subject matter Provides numerous chapter end exercises to foster conceptual understanding

Kinetic Theory of Gases Walter Kauzmann, 2013-04-22 This monograph and text was designed for first year students of physical chemistry who require further details of kinetic theory The treatment focuses chiefly on the molecular basis of important thermodynamic properties of gases including pressure temperature and thermal energy Includes numerous exercises many partially worked out and end of chapter problems 1966 edition

The Chemical News and Journal of Physical Science, 1910 *Physical Chemistry* Robert G. Mortimer, 2000-04-28 This new edition of Robert G Mortimer's Physical Chemistry has been thoroughly revised for use in a full year course in modern physical chemistry In this edition Mortimer has included recent developments in the theories of chemical reaction kinetics and molecular quantum mechanics as well as in the experimental study of extremely rapid chemical reactions While Mortimer has made substantial improvements in the selection and updating of topics he has retained the clarity of presentation the integration of description and theory and the level of rigor that made the first edition so successful Emphasizes clarity every aspect of the first edition has been examined and revised as needed to make the principles and applications of physical chemistry as clear as possible Proceeds from fundamental principles or postulates and shows how the consequences of these principles and postulates apply to the chemical and physical phenomena being studied Encourages the student not only to know the applications in physical chemistry but to understand where they come from Treats all topics relevant to undergraduate physical chemistry

Western Medical Times George Lee Servoss, 1926

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