

This review covers material that you should be familiar with before progressing to CHM1156. If you have difficulty with this material, other than just "knocking off the rust," I strongly urge you to review the early chapters in the textbook. It is imperative that you know this material so that we focus on new concepts this semester.

1. Write the empirical formula for the ionic compound formed from the following ions:
 - a. calcium and bromine
 - b. ammonium and sulfate
2. Predict whether each of the following compounds is molecular or ionic.
 - a. Ag_2SO_4
 - b. CH_3OH
 - c. $\text{Ti}(\text{NO}_3)_3$
3. Provide the chemical formulas for the following compounds:
 - a. copper(I) oxide
 - b. potassium peroxide
 - c. sulfurous acid
 - d. tetraphosphorous hexasulfide
4. Provide the proper chemical names for the following compounds:
 - a. NaHSO_3
 - b. $\text{Cr}_2(\text{CO}_3)_2$
 - c. K_2CrO_4
 - d. H_2CO_3
5. Write the balanced chemical equation for the following reaction:

Solid calcium carbide reacts with water to form an aqueous solution of calcium hydroxide and dicarbon dihydride (also known as acetylene)
6. The molecular formula of allicin, the compound responsible for the characteristic smell of garlic, is $\text{C}_6\text{H}_{12}\text{OS}_2$. What is the molecular mass of allicin? How many moles of allicin are present in 5.00 mg of this substance? How many molecules of allicin are in 5.00 mg of this substance? How many atoms of sulfur are present in 5.00 mg of this substance?
7. Rachel, a CHM1151 student, combined 30.0g of benzene and 65.0g of bromine to prepare bromobenzene: $\text{C}_6\text{H}_6 + \text{Br}_2 \rightarrow \text{C}_6\text{H}_5\text{Br} + \text{HBr}$. If her experiment produced 56.7 g of bromobenzene, what was her percent yield?
8. Alloys, or metallic mixtures, of mercury and another metal are called amalgams. Sodium in sodium amalgam reacts as follows: $2\text{Na}_{(aq)} + 2\text{H}_2\text{O}_{(l)} \rightarrow 2\text{NaOH}_{(aq)} + \text{H}_{2(g)}$. If a 15.23 g sample of sodium amalgam evolves 0.108 g of hydrogen, what is the percentage of sodium in the amalgam?
9. A compound that dissolves in water to form an aqueous solution and is an excellent conductor of electricity is always present in water as
 - a. covalent molecules
 - b. acid molecules
 - c. base molecules
 - d. salt molecules
 - e. ions

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Table of Contents Pkg Acp Cer Lab Chem 1151

1. Understanding the eBook Pkg Acp Cer Lab Chem 1151
 - The Rise of Digital Reading Pkg Acp Cer Lab Chem 1151
 - Advantages of eBooks Over Traditional Books
2. Identifying Pkg Acp Cer Lab Chem 1151
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pkg Acp Cer Lab Chem 1151
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pkg Acp Cer Lab Chem 1151
 - Personalized Recommendations
 - Pkg Acp Cer Lab Chem 1151 User Reviews and Ratings
 - Pkg Acp Cer Lab Chem 1151 and Bestseller Lists
5. Accessing Pkg Acp Cer Lab Chem 1151 Free and Paid eBooks
 - Pkg Acp Cer Lab Chem 1151 Public Domain eBooks
 - Pkg Acp Cer Lab Chem 1151 eBook Subscription Services
 - Pkg Acp Cer Lab Chem 1151 Budget-Friendly Options

6. Navigating Pkg Acp Cer Lab Chem 1151 eBook Formats
 - ePub, PDF, MOBI, and More
 - Pkg Acp Cer Lab Chem 1151 Compatibility with Devices
 - Pkg Acp Cer Lab Chem 1151 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pkg Acp Cer Lab Chem 1151
 - Highlighting and Note-Taking Pkg Acp Cer Lab Chem 1151
 - Interactive Elements Pkg Acp Cer Lab Chem 1151
8. Staying Engaged with Pkg Acp Cer Lab Chem 1151
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pkg Acp Cer Lab Chem 1151
9. Balancing eBooks and Physical Books Pkg Acp Cer Lab Chem 1151
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pkg Acp Cer Lab Chem 1151
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Pkg Acp Cer Lab Chem 1151
 - Setting Reading Goals Pkg Acp Cer Lab Chem 1151
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pkg Acp Cer Lab Chem 1151
 - Fact-Checking eBook Content of Pkg Acp Cer Lab Chem 1151
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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