

CHEM 1405 Practice Exam # 4 (Ch 8-10)

Name _____

Section # _____

MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question.

- 1) Which one of the following is characteristic of a base? 1) _____
 - A) has a sour taste
 - B) is insoluble in water
 - C) produces H_3O^+ in water
 - D) turns blue litmus red
 - E) has a slippery, soapy feel

- 2) Which of the following is correctly identified? 2) _____
 - A) HCl , weak acid
 - B) $\text{Ca}(\text{OH})_2$, weak base
 - C) NaOH , strong base
 - D) NH_3 , strong acid
 - E) H_2CO_3 , strong acid

- 3) The conjugate base of H_2O is _____. 3) _____
 - A) OH^-
 - B) O^{2-}
 - C) H_3O^+
 - D) H_2O
 - E) H_2O has no conjugate base.

- 4) Which of the following is the strongest acid? 4) _____
 - A) H_3PO_4
 - B) NH_4^+
 - C) H_2CO_3
 - D) HCl
 - E) NaOH

- 5) When a reaction is at equilibrium, _____. 5) _____
 - A) the reaction is no longer reversible
 - B) the products and reactants have the same energy content
 - C) no more reactants are converted to products
 - D) the forward and reverse reactions occur at the same rate
 - E) all reaction stops

- 6) What is the pH of a solution with $[\text{OH}^-] = 1 \times 10^{-4} \text{ M}$? 6) _____
 - A) -10.0
 - B) 10.0
 - C) 4.0
 - D) -4.0
 - E) 1.0×10^{-10}

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