

A zero order homogeneous gas phase reaction with stoichiometry, $A \rightarrow rR$, proceeds in a constant volume bomb, $\pi = 1$ when $t = 0$ and $\pi = 1.5$ when $t = 1$. If the same reaction, same feed composition and initial pressure proceeds in a constant pressure setup, find V at $t = 1$ if $V = 1$ at $t = 0$. (π = total pressure; t = time, V = volume.)

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Reaction Kinetics Volume 1 Homogeneous Gas

Shasha Hu



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