

4. Use the following data to calculate the reaction rate law for the system



- Find the order of the reactant with respect to NO and H<sub>2</sub> (2 marks)
- Calculate the value of rate constant. (1 mark)
- Write the rate law (1 mark)

Experiment	NO (mol/L)	H <sub>2</sub> (mol/L)	Initial Rate of Reaction (mol/(L·s))
1	0.001	0.004	0.002
2	0.002	0.004	0.008
3	0.003	0.004	0.018
4	0.004	0.001	0.008
5	0.004	0.002	0.016
6	0.004	0.003	0.024

# Reaction Rate Compilation For The H O N System

**Ian Pickup**



## Reaction Rate Compilation For The H O N System:

## Decoding **Reaction Rate Compilation For The H O N System**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Reaction Rate Compilation For The H O N System**," a mesmerizing literary creation penned by a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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