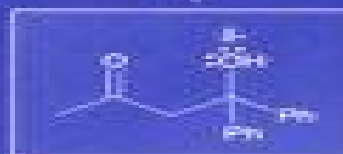
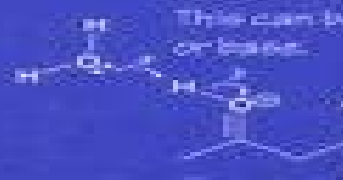


Reaction Mechanisms at a Glance

A stepwise approach to problem-solving in organic chemistry

MARK G. MOLONEY

Alcohols are nucleophiles, the oxygen possesses lone pairs, a strong acid, and fully



The intermediate product can be converted to the final product by tautomerization. This can be catalyzed by any available acid or base.

The combination NaNO_2/HCl generates HONO (nitrous acid), which is in equilibrium with N_2O_3 , a source of $\text{O}=\text{N}-\text{N}=\text{O}$. This reagent deactivates the amine functional group.



Summary: An ester reacts with 2 equivalents of PhMgBr to



Alcohols are nucleophiles and bases, since the oxygen possesses lone pairs. H^+ is a strong acid, and fully ionized ($\text{pK}_a = -10$).

Ketones are more reactive than esters, due to the stabilizing resonance delocalization in the ester, and the nitrogen on H_2N^+ . Hydroxylamine is more nucleophilic than oxygen, since nitrogen is less electronegative.

Label intermediates and reagents, and consider stereochemistry in the starting material and product.

Make the most reactive bond, if more than one exists.

Reveal the characteristic reactivity of the most reactive functional group, by considering the reaction conditions, which select the most appropriate.



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IM Harris



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