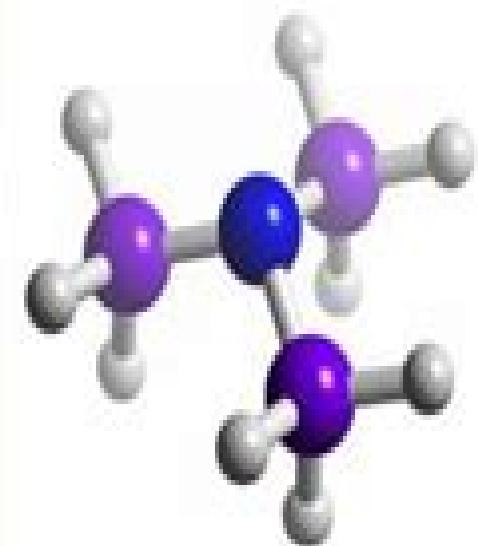


# QUÍMICA GENERAL



Dr. JAMES CELARDO NEGRETE

# Quimica General

**James Kendall**

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**Química general** Hans Rudolf Christen, 1975 El libro Química general escrito como libro de enseñanza para centros de bachillerato escuelas de comercio seminarios y otras instituciones docentes de nivel medio se publicó por primera vez en Suiza en 1963 y desde entonces se han efectuado seis ediciones sin cambios notables. Como corresponde a su título este libro da en primer lugar una visión y conocimiento generales de la materia. Porque no es precisamente un conocimiento detallado y específico de la materia lo que da un valor a la enseñanza de la Química sino un conocimiento del conjunto de leyes fundamentales y de su aplicación al mundo material de las relaciones principales en el mismo así como de los más todos y limitaciones de la investigación.

*Química general. Introducción a la Química Teórica* Cristóbal Valenzuela Calahorro, 1995-01-01

**Química general** Gordon M. Barrow, 1975-11 Sin introducir grandes modificaciones en los temas que habitualmente se encuentran en un texto de esta clase el autor ha tratado de mostrar que la Química se origina para proporcionar una perspectiva adecuada y entrar en el tema

**Química general** Ralph S. Becker, Wayne E. Wentworth, 1977 Esta Química general ha sido escrita para un curso completo de uno pensando en alumnos de Ciencias e Ingeniería. Este libro es riguroso en el sentido de que es completo. Al mismo tiempo se ha escrito a un nivel tal que pueda ser comprendido por los alumnos que inician sus estudios universitarios. No presupone conocimientos de Cálculo matemático ni de Química previamente adquiridos para su estudio.

**Química General Manual de Prácticas de Laboratorio** Instituto Tecnológico de Santo Domingo,

**Química general** Kenneth W. Whitten, Kenneth D. Gailey, 1985 Ácidos bases y sales Cinética química Compuestos de coordinación Disoluciones Electroquímica Elementos no metálicos Enlace covalente y nomenclatura inorgánica Equilibrio químico Equilibrios en disolución acuosa Estequiometría símbolos químicos fórmulas y ecuaciones Estructura atómica Los gases y la teoría cinetomolecular Líquidos y sólidos Metálicos en transición Metálicos y metalurgia Orbitales moleculares y enlace químico Periodicidad química y enlace iónico Química nuclear Química orgánica Reacciones de oxidación-reducción Reacciones químicas Reacciones químicas estudio sistemático Termodinámica química Cinética química Compuestos de coordinación Disoluciones Electroquímica Elementos no metálicos Enlace covalente y nomenclatura inorgánica Equilibrio químico Equilibrios en disolución acuosa Estequiometría símbolos químicos fórmulas y ecuaciones Estructura atómica Los gases y la teoría cinetomolecular Líquidos y sólidos

*Resolución de problemas de química general* R. Nelson Smith, Conway Pierce, 1991 Este libro será útil a aquellos alumnos que empiezan a estudiar Química y tienen dificultades al analizar los problemas y encontrar soluciones lógicas o que tienen dificultades con las representaciones gráficas y su interpretación.

**Química general** M. Garric, 1979 La novedad más importante de este curso de Química general con respecto a otros similares consiste en la introducción de los conocimientos más modernos en temas tales como 1) Estereoquímica estructuras quirales y covalentes ilustradas por más de 120 figuras 2) Teorías del enlace químico expuestas mediante algunos recursos simples de Mecánica cuántica y con numerosos ejemplos clásicos 3) Estudio de las soluciones

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**Fundamentos de la química general e inorgánica** Hans Rudolf Christen, 1977 Este texto tiene como objetivo proporcionar a los estudiantes una introducci n en los fundamentos te ricos de nuestra ciencia que les permita alcanzar simult neamente una primera versi n general sobre el campo de la Qu mica inorg nica De acuerdo con esta idea la parte general teor a at mica y de enlaces termodin mica cin tica reacciones cido base y redox Qu mica de los complejos ocupa un espacio bastante amplio mientras que la Qu mica descriptiva campo del estudio de las sustancias propiamente dicho queda m s bien en un segundo t rmico sin formar el objetivo fundamental de la obra

**Química General** James Kendall, 19??

**Química General Práctica , Prácticas de química general** Joaquín Serrano Aniorte, 1988

**Experimentación en química general** DE LA FUENTE GARCÍA SOTO, MARÍA DEL MAR,DÍAZ LORENTE, VÍCTOR MANUEL,MARTÍNEZ URREAGA, JOAQUÍN,NARROS SIERRA, ADOLFO,POZAS REQUEJO, FRUTOS,2006-01-01 En esta obra se recogen materiales para un curso de Experimentacion en Quimica General esto es para un primer curso de experimentacion o para la parte experimental de las asignaturas de Quimica de los primeros cursos universitarios de titulaciones como Quimica Fisica Biologia FarmaciaB Entre los contenidos se incluyen cuestiones generales como seguridad toma y tratamiento de datos o presentacion de resultados una colección de practicas de diferente nivel de dificultad cuidadosamente elegidas en funcion de su accesibilidad precio seguridad e interes para conseguir los objetivos de un curso de este tipo y finalmente algunos ejemplos de pequenos proyectos para ser realizados por los alumnos de forma mas autonoma

**Química general: manual de ejercicios y problemas, primera parte**,1980

**Lecciones elementales de química general** Ramon Torres Muñoz de Luna,1872

**Manual pedagógico de prácticas de química general en microescala** Beatriz Virginia Cervantes Nemer,2006

**Temas de Química General** Niko Hilje Quirós,2004

**Lecciones elementales de química general** D.R.T. MUÑOZ DE LUNA,1864

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